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1. Which ones of the following reactions are endothermic in other words ΔH is positive? I. $H_2O(l) + 10,5kcal \rightarrow H_2O(g)$ ΔH 1. II. $2NH_3 + 22kcal \rightarrow N_2 + 3H_2$ ΔH 2. III. $Na + Energy \rightarrow Na^+ + e^-$ ΔH 3. Solution:

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Ch 17 Thermochemistry Practice Test Matching Match each item with the correct statement below.

a. calorimeter d. enthalpy b. calorie e. specific heat c. joule f. heat capacity ____ 1. quantity of heat needed to raise the temperature of 1 g of water by $1^\circ C$ ____ 2. SI unit of energy ____ 3.

Ch 17 Thermochemistry Practice Test

Thermochemistry Test Preview Matching Match each item with the correct statement below. a.

calorimeter d. enthalpy b. calorie e. specific heat c. joule f. heat capacity ____ 1. quantity of heat needed to raise the temperature of 1 g of water by $1^\circ C$ ____ 2. ____ SI unit of energy 3. ____ quantity of heat needed to change the temperature of 1 g of a ...

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Ch 17 Thermochemistry Practice Test Matching Match each item with the correct statement below.

a. calorimeter d. enthalpy b. calorie e. specific heat c. joule f. heat capacity ____ 1. quantity of heat needed to raise the temperature of 1 g of water by $1^\circ C$ ____ 2.

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AP Chemistry Practice Test, Ch. 6: Thermochemistry Name ____ MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. 1) A chemical reaction that absorbs heat from the surroundings is said to be ____ and has a ____ ΔH at constant pressure. A)endothermic, positive

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A.P. Chemistry Practice Test: Ch. 11, Solutions Name_____ MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. 1) Formation of solutions where the process is endothermic can be spontaneous provided that _____. A)the solvent is a gas and the solute is a solid

A.P. Chemistry Practice Test: Ch. 11, Solutions MULTIPLE ...

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Question & Answers (0) With each cycle, a 2650 W engine extracts 2330 J from a thermal reservoir at 85.1 C and expels 1840 J into a thermal reservoir at 20.9 degree C. a) What is the work done for ...

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Science Quiz: Chemistry: Solutions

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THERMOCHEMISTRY - Heat Of Reaction And Heat Of A S ...

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Thermochemistry Study Guide. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. shelby_martell. Terms in this set (27) Energy. the ability to do work or produce heat. Potential Energy. in chemistry this is usually the energy stored in bond. Kinetic Energy.

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162 CHAPTER 6: THERMOCHEMISTRY To convert the answer to joules, we write: $101.3 \text{ J } 0.18 \text{ L atm } 1 \text{ L atm} = - \cdot \times = \cdot w -18 \text{ J}$ 6.17 An expansion implies an increase in volume, therefore w must be -325 J (see the defining equation for pressure-volume work.) If the system absorbs heat, q must be $+127 \text{ J}$. The change in energy (internal

CHAPTER 6 THERMOCHEMISTRY

A solution is made by dissolving 3.35 g of an unknown salt in 150 mL of water, and the temperature of the water rises 3.0°C. The addition of a silver nitrate solution results in a precipitate. Assuming that the heat capacity of the solution is the same as that of pure water, use the information in Table

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9.5.1 and solubility rules to identify ...

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