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neutral group of atoms
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covalent bonds.
Diatomic Molecule. a
molecule consisting of
two atoms. Molecular
Compound.

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A covalent bond in which the bonding electrons are most likely to be found in sausage-shaped regions above and below the bond axis of the bonded atoms.

tetrahedral angle A bond angle of 109.5 degrees that results when a central atom forms four bonds

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directed toward the center of a regular tetrahedron.

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bonds. a molecule consisting of two atoms. the mnemonic for remembering which non-metal elements form dia....

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Covalent Bonding
Single Covalent Bonds
When only one pair of electrons is shared, such as in a hydrogen

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molecule, it is a single covalent bond. The shared electron pair is often referred to as the bonding pair. For a hydrogen molecule, shown in Figure 8.4, each covalently bonded atom equally attracts the pair of shared electrons.

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8.1 Molecular
Compounds > 21

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molecular structure of
water shows how the
oxygen ...

**How are atoms
joined together to
make compounds
with ...**

8.2 The Nature of

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Experimental evidence,
however, indicates ...

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Bonding In ionic
bonding, atoms
transfer electrons to
achieve noble gas
configuration. In
covalent bonding,

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atoms share electrons to achieve noble gas configuration. Most atoms share electrons until they have a total of 8 valence

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often referred to as the bonding pair. For a hydrogen molecule, shown in Figure 8.4, each covalently bonded atom equally attracts the pair of shared electrons.

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the essential lessons
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atoms in a bond provides the pair of bonding electrons; this is a 7. is required to break covalent bonds between 8.

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B to the correct term in
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single covalent bond a.
a chemical formula
that shows the

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arrangement of atoms
in a molecule or a
polyatomic ion

b. HCN - SharpSchool

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