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Solution: 2.33. 19.5g of CH₂FCOOH is dissolved in 500g of water. The depression in the freezing point of water observed is 1.0°C. Calculate the van's Hoff factor and dissociation constant of fluoroacetic acid. Solution: 2.34. Vapour pressure of water at 293 K is 17.535 mm Hg.

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Furthermore, the chapter involves the properties of the solutions. This chapter has information on types of solutions and their formation. Most noteworthy, the chapter deals with various alternatives in which concentrations can be expressed. Subtopics covered under NCERT Solutions for Class 12 Chemistry Chapter 2. 2.1 Types of Solutions; 2.2 ...

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