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## **Chapter 15 Water And Aqueous**

Chapter 15 Water and  
Aqueous Systems159  
SECTION 15.1 WATER

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Chapter 15 Water

AND ITS PROPERTIES  
(pages 445–449) This section describes the properties of water in the liquid and solid states and explains how hydrogen bonding affects the surface tension and vapor pressure of water.

Water in the Liquid State (pages 445–447)

1.

**SECTION 15.1**  
**WATER AND ITS**  
**PROPERTIES (pages**

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Aqueous Systems".  
The Water Molecule: a  
Review. • Water is a  
simple tri-atomic  
molecule, H. 2. O.

- Each O-H bond is highly polar, because of the high electronegativity of the oxygen (N, O, F, and Cl have high values)
- bond angle of water = 105°.

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Chemistry, Chapter 15,  
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Systems. surface  
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tension. surfactant.

aqueous solution.

solvent. the inward

force or pull that tends  
to minimize the surface

ar.... any substance  
that interferes with

hydrogen bonding

between wa.... is water  
that contains dissolved  
substances.

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yourself study the  
vocabulary and major  
concepts presented in  
this chapter.

heterogeneous mixture  
containing particles  
that are small enough  
to remain dispersed in  
the solvent and do not  
separate on standing.

## **Quia - Chapter 15 "Water and Aqueous Systems"**

Section 15.1 - Water  
and its Properties: File  
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1. Chapter 15 Chapter  
17 in your books“Water  
and Aqueous Systems”

2. Section 15.1 Water  
and its

PropertiesOBJECTIVES:

- Explain the high surface tension and low vapor pressure of water in terms of the structure of the water molecule and hydrogen bonding. •

**Chemistry - Chp 15 -**

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Chapter 15 Water  
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**Systems - Notes**

Chapter 15 - Water and  
Aqueous Systems -

15.3 Heterogeneous  
Aqueous Systems -

15.3 Lesson Check -

Page 507: 21 Answer

Colloids cannot be separated through filtering because their particles are smaller than those of suspensions and they do not settle out over time.

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existing beneficial facts  
about different subject  
matter plus topics.

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Water and Aqueous  
Systems. Pre-AP  
Chemistry ; Charles  
Page High School ;  
Stephen L. Cotton; 2  
Chapter 15 Review.  
Surface tension is the  
\_\_\_\_. How does the

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surface tension of  
water compare with  
the surface tensions of  
most other liquids?

Which type of  
mixture(s) exhibit the  
...

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OBJECTIVES: -Explain the high surface tension and low vapor pressure of water in terms of the structure of the water molecule and hydrogen bonding.

3.

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friend; Find other activities ... a relatively strong intermolecular force responsible for water's high surface tension: suspension: a mixture from which particles settle out upon standing: aqueous solution: a solution in which the solvent is water: solvent: the ...

### **Quia - Chapter 15**

### **"Water and Aqueous Systems"**

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Chapter 15 Water

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Chapter 15 - Water and  
Aqueous Systems -

15.1 Water and Its  
Properties - 15.1

Lesson Check - Page

493: 2 Answer The

structure of ice is a

regular open

framework of water

molecules arranged

like a honeycomb.

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SECTION 15.1 WATER  
AND ITS PROPERTIES

(pages 445–449)

CHAPTER 15, Water and  
Aqueous

Systems(continued) 6

Circle the letter next to  
each sentence that  
describes a result of  
the surface tension of

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water a In a full glass of water, the water surface seems to bulge over the rim of the glass b Water beads up into

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